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rings. • Common

Characteristics:

1. Generally non-
polar (won ' t
dissolve in water)

2. Soluble in non-

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polar solvents

3. Non-electrolytes.

4. Low melting
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Alkanes have only a single bond in its molecular formula.

A hydrocarbon molecule is said to be saturated if there are no double or triple bonds.

Alkenes have one or more than one double bond.

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Answer true or
false: Alkanes are
saturated
hydrocarbons ...

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ChemFacts. • The
primary source of
hydrocarbons is
petroleum. • About
75 million barrels of
petroleum are

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pumped out of the Earth each day. • Hydrocarbons are used as fuels and are the raw materials for products such as plastics, synthetic fibers, solvents, and industrial chemicals.

Chapter 21:
Hydrocarbons

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carbon and hydrogen. Common hydrocarbons include methane, propane, and butane. Section 22 hydrocarbon Compound Answer

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Because carbon has
four valence
electrons, car-bon
atoms always form

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four covalent bonds.

- The carbon atoms in an alkane can be arranged in a straight chain or in a chain that has branches.

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products such as
plastics, synthetic
fibers, solvents, and
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chemicals.

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carbon and hydrogen. Common hydrocarbons include methane, propane, and butane.

Hydrocarbons are the primary constituents of fossil...

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alkanes If all of the
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by which
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natural gas This is
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80% methane, 10%

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2. Soluble in non-polar solvents
3. Non-electrolytes.
4. Low melting

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