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and Measurement~~

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matter - Part-1 -
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Chemical and Physical
Changes

States of Matter : Solid
Liquid Gas 10 Amazing
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Pure Substances and
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Understanding Atoms,
elements, and molecules
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Chemicals Ozone
Layer, atmosphere,
ozone formation,
chlorofluorocarbons,
CFC ' s 1.2 Chemistry
and Matter Chemistry
Central Science

Chapter Summaries –
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Chemistry Matter and
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Chapter 1 Section 2

MATTER and Its

Properties Matter –

Anything that has

MASS and VOLUME.

ATOMS and

MOLECULES the

basic building blocks of

matter. ATOM-- the

smallest unit of an

element that still

maintains the properties

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that element.

ELEMENT — a pure substance made up of only one type of atom.

**MATTER and
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Section 1**

**Chemistry: Matter and
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of the choice that best

completes the statement

or answers the question.

1. Atomic radii cannot

be measured ...

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CHAPTER 3

SOLUTIONS

MANUAL

Matter—Properties and
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Solutions Manual

Chemistry: Matter and
Change • Chapter 3

35 Section 3.1

Properties of Matter

pages 70 – 75 Problem-

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Solving Lab 1. Explain why the flow of a compressed gas must be controlled for practical and safe use.

Matter—Properties and Changes
Matter—Properties and Changes

Chapter 1

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Guide - Chapter 6 –

The Periodic Table and

Periodic Law Section

6.1 Development of the

Modern Periodic Table

1. octaves 2. eight 3.

nine 4. accepted 5.

Dmitri Mendeleev 6.

atomic mass 7. elements

8. atomic number 9.

Henry Moseley 10.

protons 11. periodic law

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12. properties 13. 14 ...

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in Atoms Reviewing
Vocabulary Match the
definition in Column A
with the term in
Column B. Column A
Column B 1. The set of

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frequencies of the
electromagnetic waves
emitted by the atoms of
an element 2. The
minimum amount of
energy that can be lost
or gained by ...

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changes (that is, energy produced or absor.... A system that can exchange neither energy nor matter with its...

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extensive property.

intensive property.

extensive properties

(examples) intensive

property (examples) a

property that depends

on the amount of matter

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in a sample. a property that depends on the type of matter in a sample not.... mass, volume, length, weight.

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31 number of moles of

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known substance C H C

H C H O O O CO CO

CO H O H O H O

number of moles of

unknown substance O

CO H O C H CO H O

C H O H O C H O CO

use mole ratio 5 mol O

2 1 mol C 3 H 8 3 mol

CO 2 1 mol C 3 H 8 4

mol H 2 O 1 mol C 3 H

8 1 mol C 3 H 8 5 mol

O 2

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MATH SKILLS TRANSPARENCY MASTER 16

A “chemical change” is a change that produces matter with a different composition than the original matter. Chemical Change A change in which one or more substances are converted into different substances. Heat and light are often evidence

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of a chemical change.

Properties of
Compounds Quite
different properties than
their component
elements.

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